## CO<sub>2</sub> Capture with Ionic Liquids Involving Phase Change

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#### **Technology Summary**

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A new concept for  $CO_2$  capture that uses *phase change ionic liquids (PCILs)* offers the potential to significantly reduce parasitic energy losses incurred from capturing  $CO_2$  from flue gas. PCILs are solid ionic materials that have high  $CO_2$  uptake (one mole of  $CO_2$  for every mole of salt at post-combustion flue gas conditions) and form a liquid when they react with  $CO_2$ . This allows for a novel process that uses the heat of fusion to provide part of the heat needed to release  $CO_2$  from the absorbent, reducing the total energy required. This project will (1) develop and characterize PCILs; (2) evaluate energy savings in a new  $CO_2$  capture process; and (3) demonstrate the technology at laboratory scale.





Goal: develop ionic salts that undergo a phase change (from solid to liquid) when they react with  $CO_2$ ; taking advantage of the enthalpy change when PCILs react with  $CO_2$  to enable capture of 90% of the  $CO_2$  from post-combustion flue gas with less than a 35% increase in the cost of electricity.



In a 500 MW (471 MW de-rated) coal plant:

- •Aqueous amine scrubbing incurs parasitic energy losses of 28% (132 MW).
- •Current ionic liquids could reduce this to 23% (110 MW)

### **Progress to Date**

Synthesized five Gen1 PCILs

process

- Measured CO<sub>2</sub> uptake of two Gen1 PCILs and began measurements of other compounds
- Developed forcefields for Gen1 PCILs and initiated molecular simulations
- Initiated measurements of heats of reaction and construction of packed bed absorption column
- Initiated process modeling







#### **University of Notre Dame Team**

